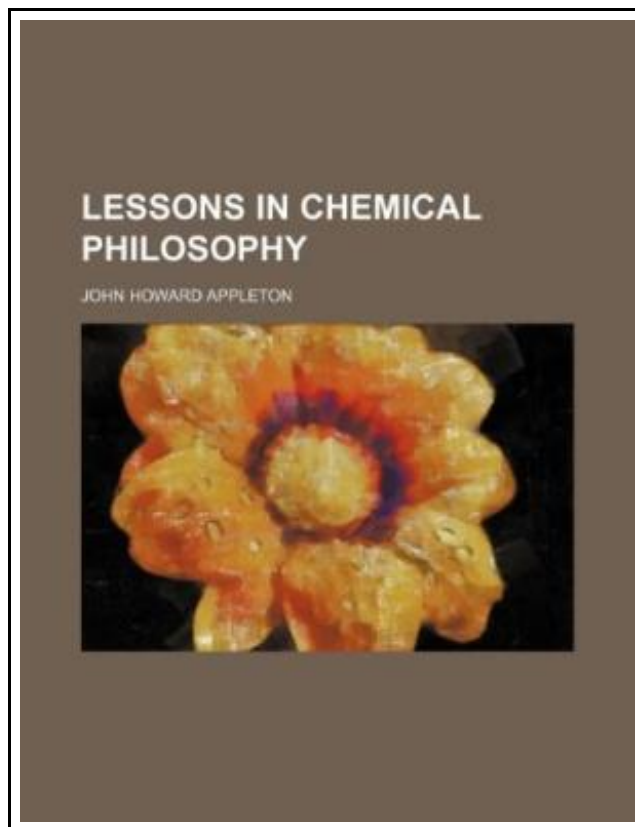


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Reviews

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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1890 Excerpt: .is remarkable for absorbing or occluding, at ordinary temperature, eight hundred times its bulk of hydrogen gas. The late Professor Graham, of London, who observed this property of palladium, considered the solid thus formed to be an alloy, and to contain hydrogen in the solid form. The quantities of the two elements are in this case approximately in the proportion of the weight of one atom of hydrogen to one atom of palladium, so that it has been suggested that the substances may be in chemical combination. The metal platinum has the same power as palladium, though to a less degree. If a warm piece of platinum foil is placed in a current of mixed illuminating gas and air, the foil absorbs portions of all the gases. In so doing it condenses them to such a degree as to bring the molecules very near to each other, even within that minute distance through which chemical 1 Section (B) is at p. 114. attraction can be exerted. Chemical union does in fact take place, as is evidenced by the production of light and heat and other phenomena of true combustion. Hannay and Hogarth have shown that in some cases a gas, brought in contact with a solid, dissolves the latter quickly into itself. j Fig. 87.--Diibereiner s lamp, showing the adhesion of hydrogen gas to platinum. The bottom of the lamp is a hydrogen generator. Dilute sulphuric acid acts upon the mass of zinc B, Hydrogen rises...



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